## SNC 2P Exam Review

## Terms to Know

| Corrosive |  |
| :--- | :--- |
| Compound |  |
| Atom |  |
| Physical Change |  |
| Law of Conservation of Mass |  |
| Alkali Metals |  |
| Halogens |  |
| Cell |  |
| Carfusion |  |

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| Tissue |  |
| :--- | :--- |
| Peristalsis |  |
| Angle of Reflection |  |
| Bioluminescence |  |
| Triboluminescence |  |
| Incandescence |  |
| Chemiluminescence |  |
| Refraction |  |

## Biology:

List the cell organelles and briefly outline their function:

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Contrast animal and plant cells.

Briefly outline the stages of mitosis.

Why do cells divide?

List and describe some ways to combat cancer?

What are the major components of the circulatory system?

What are the major components of the respiratory system?

What are the major components of the digestive system? Be sure to include function.

What are the two types of digestion?

## Chemistry:

List the charges of the following ions: Flouride ion, Chloride ion, Sodium ion

Draw a Bohr-Rutherford diagram of a helium atom.

Name the following ionic compounds:
$\mathrm{Na}_{2} \mathrm{O}$

KCl $\qquad$

Name the following molecular compounds:
$\mathrm{CH}_{4}$ $\qquad$
$\mathrm{S}_{2} \mathrm{O}_{2}$ $\qquad$
How many atoms are in $\mathrm{Mg}(\mathrm{OH})_{2}$ ?

Balance the following chemical reactions:

$$
\begin{array}{lllll}
\mathrm{Li} & + & \mathrm{AlF}_{3} \rightarrow & \mathrm{Al} & + \\
\mathrm{NiFl} \\
\mathrm{NaCl} & \mathrm{LiBr} & \rightarrow & \mathrm{NaBr} & + \\
\mathrm{LiCl} \\
\mathrm{H}_{2}+ & \mathrm{O}_{2} & \rightarrow & \mathrm{H}_{2} \mathrm{O} &
\end{array}
$$

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Give an example of the following:
A synthesis reaction
A decomposition reaction
A double displacement reaction
A single displacement reaction

What are some characteristics of acids and bases?

## Physics:

What is the difference between a concave and a convex mirror?

Describe the image that plane mirrors produce.

Locate the image in the following lens
$\qquad$

L
T $\qquad$

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Locate the image in the following mirror


