

Naming Compounds

sodium hydroxide _____

CaBr_2 _____

lithium oxide _____

$(\text{NH}_4)_2\text{SO}_4$ _____

cobalt (III) carbonate _____

NaNO_3 _____

aluminum sulfide _____

CCl_4 _____

iron (III) phosphide _____

CuO _____

sodium permanganate _____

SO_3 _____

manganese (III) fluoride _____

AgNO_3 _____

beryllium nitrate _____

Zn_3N_2 _____

nickel (III) sulfite _____

OF_2 _____

potassium oxide _____

$\text{Al}_2(\text{CO}_3)_3$ _____

silver bromide _____

Pb_3N_4 _____

zinc phosphate _____

NH_3 _____

dicarbon tetrahydride _____

CO_2 _____

tin (II) hydroxide _____

Ag_2SO_4 _____

Classifying and Balancing Reactions

Classify each reaction as synthesis, decomposition, single displacement, double displacement or combustion. Then balance them.



Type: _____



Type: _____



Type: _____



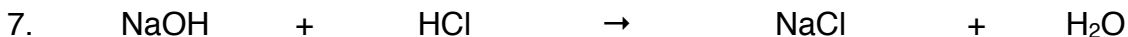
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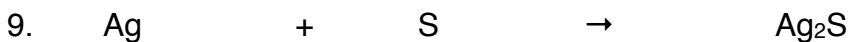
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