

1)Identify the number of carbons.

2) Use the appropriate IUPAC prefix with the ending _____.

1)Draw the numbe	er of	
identified by the IUPAC prefix. Attach		
them each by a	bond.	
2)	each carbon using	
a		



Examples	Examples
Name: CH ₃ - CH ₂ - CH ₃	Draw: pentane
CH3 - CH2 - CH2 - CH2 - CH2 - CH3	octane
Alkanes are characterized by a Alkanes are and Alkanes always end with ''	carbon-carbon bond. d contained no double or triple bonds.

* Fill in the Blanks Here

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EXAMPLES

Name: CH₃ - CH₂ - CH₃

CH3 - CH2 - CH2 - CH2 - CH2 - CH3

octane

pentane

EXAMPLES

• Alkanes are characterized by a _____ carbon-carbon bond.

Alkanes are ______ and contained no double or triple bonds.

Alkanes always end with '____





In ketones, the double bonded oxygen is always in the MIDDLE



U



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Naming Ketones

- * Identify the number of carbons.
- * Use the appropriate IUPAC prefix with the ending ONE.
- * State the location of the oxygen using a number.
- * Name any additional side chains with the same numbering system.



$CH_3 - CH_2 - CH_3$

0



0

I



$CH_3 - CH_2 - CH_2 - CH_2 - CH_2 - C - CH_2 - CH_2 - CH_2 - CH_3$

0

CH3

Decan-5-one

$CH_2 - CH_2 - C - CH_2 - CH_2 - CH - CH_3$

 \mathbf{O}

6-methylheptan-3-one

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Examples	Examples
Name:	Draw:
CH ₃ - CH ₂ - CH ₃	pentane
CH3 - CH2 - CH2 - CH2 - CH2 - CH3	octane
Alkanes are characterized by a	carbon-carbon bond.
Alkanes are and contained no double or triple bonds.	

Alkanes always end with '_____

Drawing Ketones

- * Praw the number of carbons identified by the IUPAC prefix. Attach them each by a single bond.
- * Praw the double bonded oxygen as identified.
- * Add any additional side chains.
- * Saturate each carbon using a hydrogen.



3-propyloctan-2-one



3-propyloctan-2-one

 $\begin{array}{c} CH_3 - CH - CH - CH_2 - CH_2 - CH_2 - CH_2 - CH_3 \\ II I \\ 0 \\ CH_2 - CH_2 - CH_3 \end{array}$