

# Carboxylic Acids

## Carboxylic Acids

- A ketone is a hydrocarbon derivative that contains a **carboxyl** functional group at the **end** on the base chain.

## IUPAC Names

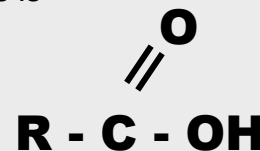
5C- pentanoic acid

9 C- nonanoic acid

4C- butanoic acid

10 C- decanoic acid

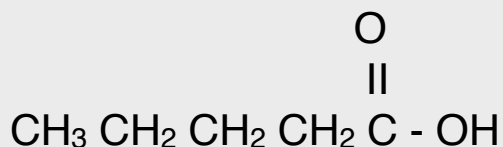
- The general formula for the series is



## Naming Carboxylic Acids

- 1) Identify the base number of carbons.
  - The base is the longest continuous chain of carbons that contains the carboxyl group.
  - Use the number of carbons as a prefix before the suffix **-oic acid**
- 2) Since the carboxyl group is always at the end, it's location doesn't have to be provided. The carbon that contains the carboxyl group is always carbon 1.
- 3) Name any additional side chains and identify their positions using numbers.

Example:



Answer: pentanoic acid

## Drawing Carboxylic Acids

- 1) Start by drawing the base chain. Draw the number of carbons as indicated by the prefix.
- 2) Add the carboxyl group to the appropriate carbon.
- 3) Add any indicated double or triple bonds.
- 3) Add any indicated side chains.
- 4) Saturated the remaining carbons

Example: 3-methylbutanoic acid

Answer:

